

MSE-441

Electrochemistry for materials technology

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Cursus	Sem.	Type
Materials Science and Engineering	MA1, MA3	Opt.

Language of teaching	English
Credits	2
Session	Winter
Semester	Fall
Exam	During the semester
Workload	60h
Weeks	14
Hours	2 weekly
Courses	2 weekly
Number of positions	

Summary

This course aims at familiarizing the student with state of the art applications of electrochemistry in materials science and technology as well as material requirements for electrochemical engineering.

Content

The course includes a revision of the basic concepts of electrochemistry and of the electrochemical techniques followed by the discussion of relevant applications for surface modifications (galvanic coatings technology, surface structuration, micro/nano fabrication) and energy issues (materials for batteries, fuel cells, hydrogen generation) as well materials aspects in electrochemical engineering (catalytic electrodes, analytical electrochemistry).

Keywords

Materials, Electrochemistry, Micro-fabrication, Coatings, Energy generation, Energy conversion

Learning Prerequisites**Required courses**

Chimie générale, Introduction à la science des matériaux

Recommended courses

Métaux et alliages

Important concepts to start the course

General chemistry: thermodynamics, kinetics, equilibrium, acid-base and complexation reactions, redox reaction.

Metallurgy: microstructure of metals and alloys, mechanical properties, deformation and rupture.

Physics: electrical circuits, transport phenomena

Learning Outcomes

By the end of the course, the student must be able to:

- Use electrochemical concepts and methods for materials science
- Design micro/nano materials via electrochemical processes
- Structure surfaces with tailored properties
- Design appropriate materials for electrochemical systems
- Analyze electrochemical processes and devices

- Manage electrochemical material fabrication
- Describe electrochemical reactions
- Formulate requirements for energy generation and storage materials

Teaching methods

Ex cathedra with excercises and case studies.

Expected student activities

Active participation during lectures and in the resolution of excercies, group work in case studies

Assessment methods

Oral presentation

Supervision

Office hours	No
Assistants	No
Forum	No
Others	Meetings with teacher upon appointment establised by email

Resources

Bibliography

Electrochemsity theory: D. Landolt, Corrosion and Surface Chemistry of Metals, CRC/EPFL Press 2007

Ressources en bibliothèque

- [Corrosion and surface chemistry of metals / Landolt](#)

Notes/Handbook

Copy of slides available fromt eh website

Websites

- <http://tic.epfl.ch>