

CH-311

**Macromolecular structure and interactions**

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Cursus	Sem.	Type
Chemistry	BA5	Obl.
HES - CGC	H	Obl.

Language of teaching	English
Credits	2
Session	Winter
Semester	Fall
Exam	Written
Workload	60h
Weeks	14
<b>Hours</b>	<b>2 weekly</b>
Courses	2 weekly
<b>Number of positions</b>	

**Summary**

This course covers the basic biophysical principles governing the thermodynamic and kinetic properties of biomacromolecules involved in chemical processes of life. The course is held in English.

**Content**

- **The conformation of biological macromolecules and membranes**
  - Forces in biomolecules
  - Protein primary and secondary structure
  - Tertiary structure of proteins
  - DNA structure
  - Conformations of unstructured polymers in solution (Gaussian chain models, freely-jointed chain, wormlike chain)
- **Conformational equilibria and dynamics of polypeptides and proteins**
  - Thermodynamics of protein folding (folding equilibria, calorimetry of protein folding transitions)
  - Kinetics of protein folding (folding pathways, intermediates)
  - Conformational transitions in proteins (native state fluctuations, allostery, structural rearrangements in enzyme catalysis)
  - Thermodynamics and kinetics of alpha-helix - coil transition
- **Spectroscopy of Biomolecules**
  - Biomolecular absorption spectroscopy (UV absorption, circular dichroism)
  - Biomolecular fluorescence
  - Structural biology: X-ray crystallography, electron microscopy and NMR spectroscopy of proteins
- **Ligand-receptor interactions**
  - Equilibrium binding reactions
  - Binding inhibition
- **Transport phenomena and stochastic processes in biology**
  - Fluctuations in biology
  - Macromolecular diffusion

**Keywords**

biophysics, biophysical chemistry, protein, nucleic acid, structure, thermodynamics, kinetics, protein folding, spectroscopy, fluorescence, absorption, helix-coil, fluctuations, receptor, ligand

## Learning Prerequisites

### Required courses

Biochemistry I  
Chemical thermodynamics

### Important concepts to start the course

General chemical and biochemical concepts

## Learning Outcomes

By the end of the course, the student must be able to:

- Describe physical chemistry methods in biology
- Integrate chemical and physical concepts in biology
- Reason which methods are appropriate for a biological problem

## Teaching methods

ex cathedra

## Expected student activities

Attendance of the lectures  
Study of indicated materials

## Assessment methods

Written exam

## Resources

### Bibliography

"Principles of Physical Biochemistry", Van Holde, Prentice Hall  
"Physical Biology of the Cell", Phillips, Kondev, Theriot, Garcia, Garland Science

### Ressources en bibliothèque

- [Physical biology of the cell / Phillips](#)
- [Principles of physical biochemistry/ Van Holde](#)

## Prerequisite for

Dynamics of biomolecular processes  
Chemical biology  
Experimental biological & biophysical chemistry  
Nanobiotechnology and Biophysics  
Cellular Signaling